

Marcus Theory of Electron Transfer

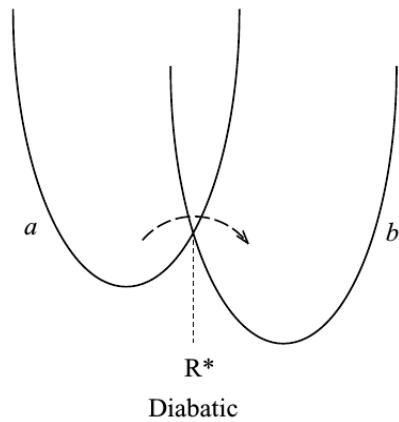


The Nobel Prize in Chemistry 1992

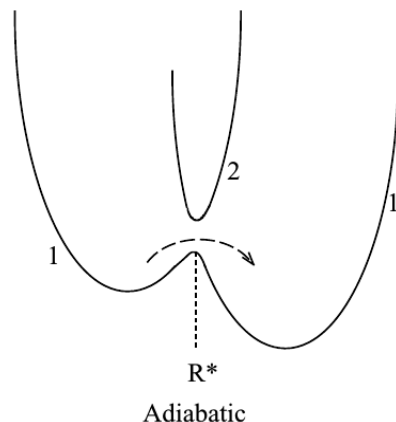
for his contributions to the theory of electron transfer reactions in chemical systems; however the full scope of Marcus Theory is much broader and can be used in all condensed matter.

$$k_{ET} = Z_{ab} \exp \left\{ -\frac{(\Delta G_{ab}^0 + \lambda)^2}{4\lambda k_B T} \right\}$$

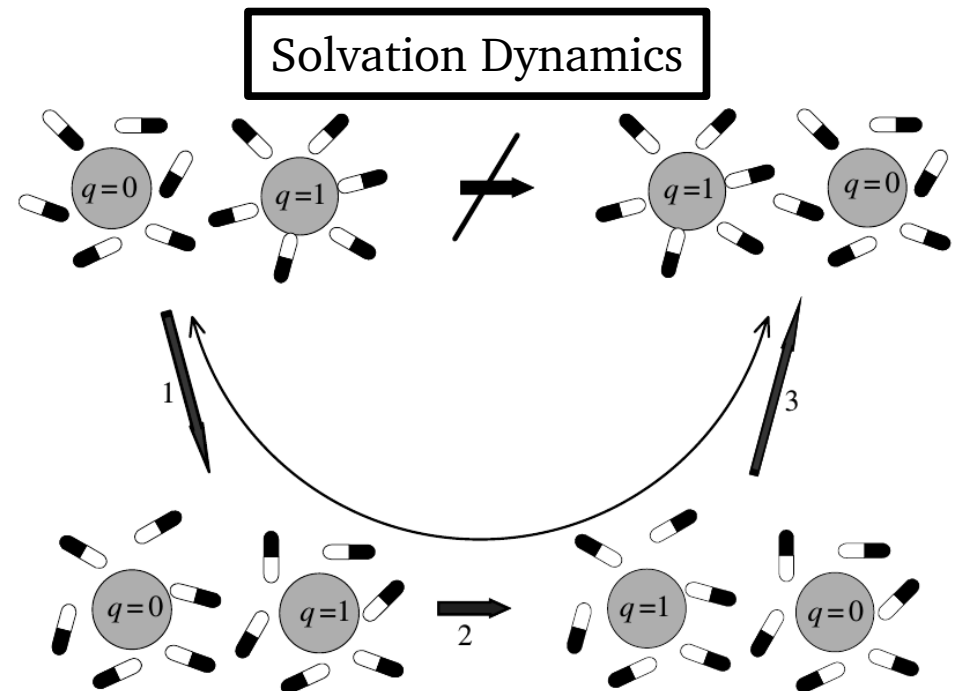
Rudolph A. Marcus



weak electronic coupling



strong electronic coupling



adapted from *Chemical Dynamics in Condensed Phases*, A. Nitzan, Oxford Press.

Marcus Theory of Electron Transfer

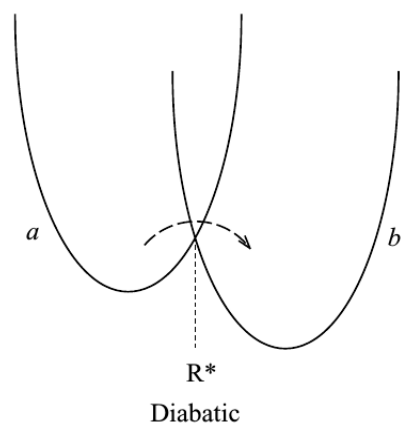


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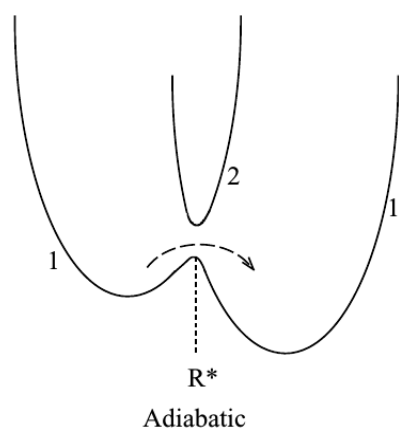
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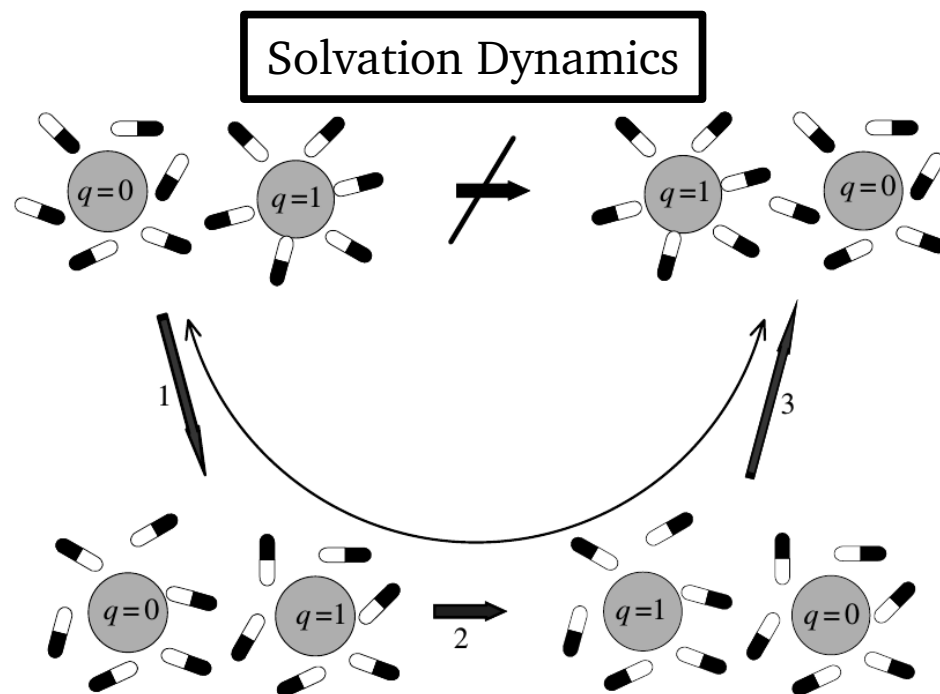
$$k_{ET} = \frac{2\pi}{\hbar} |V_{ab}|^2 \frac{1}{\sqrt{4\pi\lambda k_B T}} \exp \left\{ -\frac{(\Delta G_{ab}^0 + \lambda)^2}{4\lambda k_B T} \right\}$$



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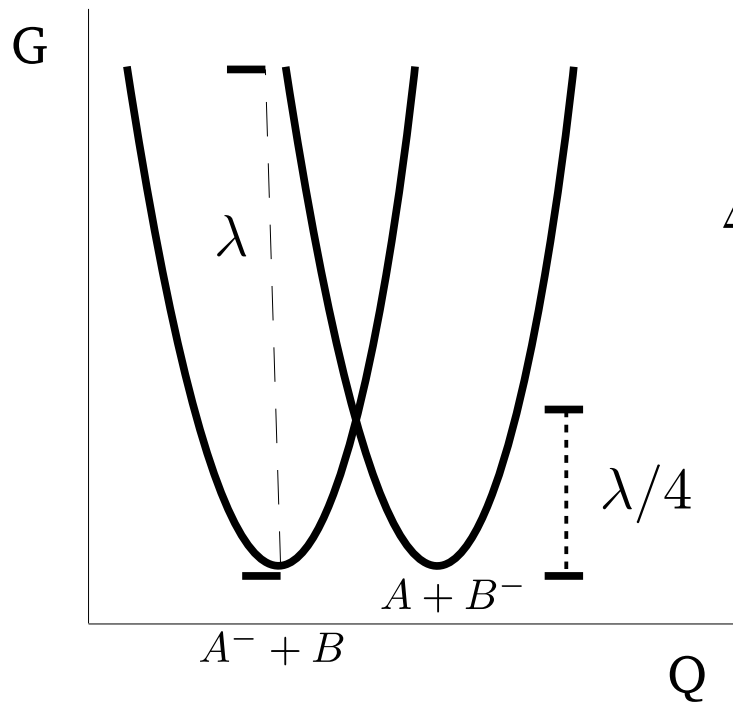
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$$\frac{2\pi}{\hbar} |V_{ab}|^2 = \text{collision frequency} \sim 10^{-9} - 10^{-10} \text{ sec}^{-1}$$

$$\Delta G_{ab}^0 = \text{free energy change for the ET}$$

The Marcus Parabolae:



$$\Delta G^0 = 0 \implies \exp \left\{ -\frac{(\Delta G_{ab}^0 + \lambda)^2}{4\lambda k_B T} \right\} = e^{-\frac{\lambda/4}{k_B T}}$$

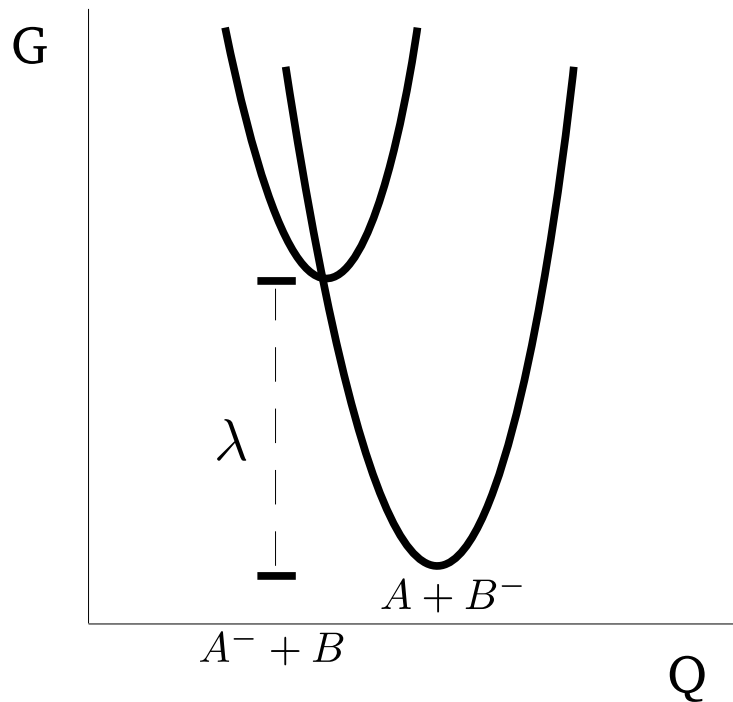
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The Marcus Parabolae:



exergonic (spontaneous) reaction:

$$\Delta G = G_{\text{products}} - G_{\text{reactants}} < 0$$

$$\Delta G_{ab}^0 = -\lambda$$

$$\Rightarrow \exp \left\{ -\frac{(\Delta G_{ab}^0 + \lambda)^2}{4\lambda k_B T} \right\} = e^{-\frac{0}{k_B T}} = 1$$

diffusion controlled

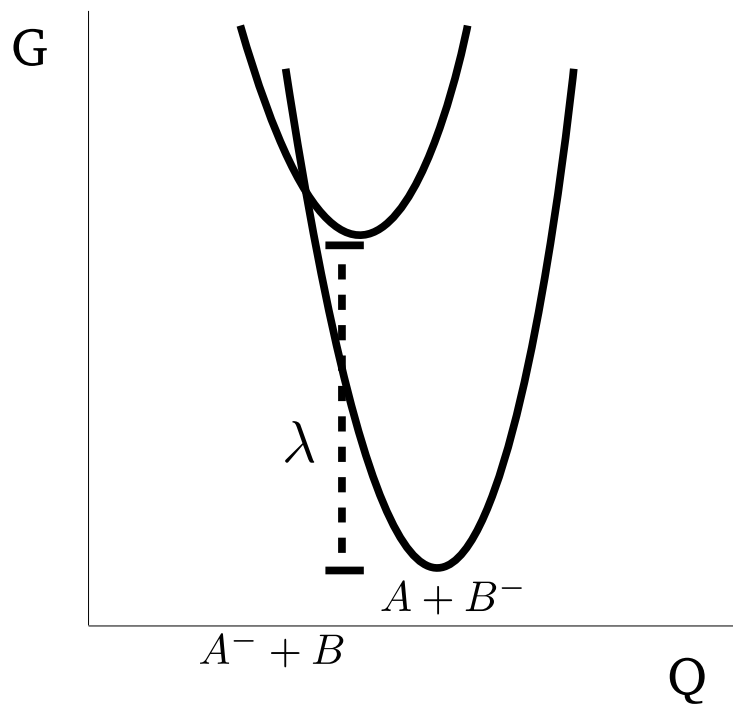
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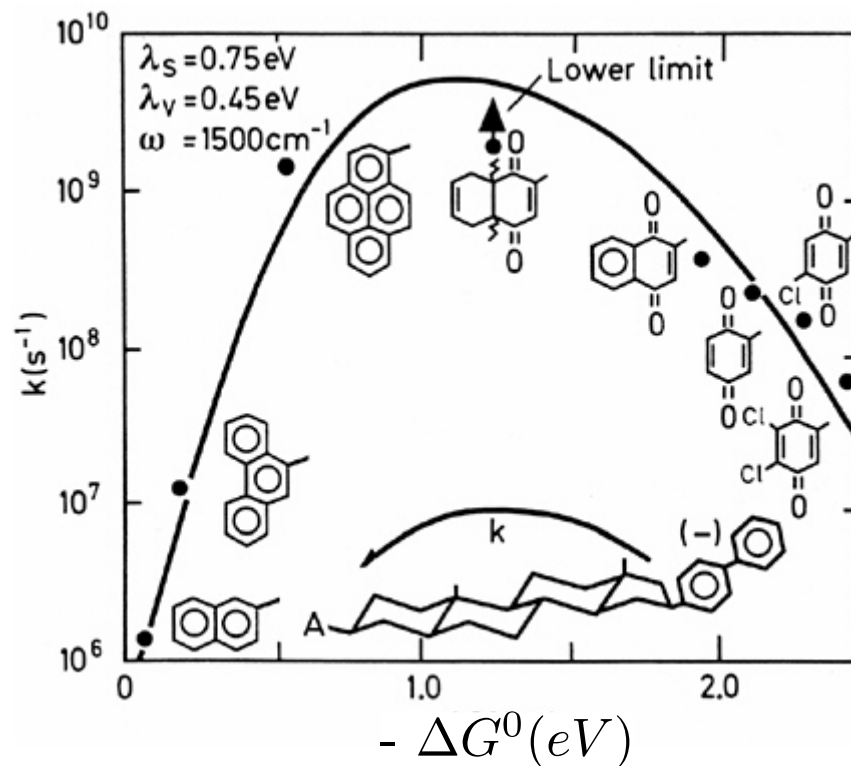
$$|\Delta G_{ab}^0| > \lambda$$

$$\Rightarrow \exp \left\{ -\frac{(\Delta G_{ab}^0 + \lambda)^2}{4\lambda k_B T} \right\} < 1$$

Inverted region: $k_{ET} < k_{max}$

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G. L. Closs and J. R. Miller. "Intramolecular long-distance electron transfer in organic molecules". *Science* 240(1988):440-47.

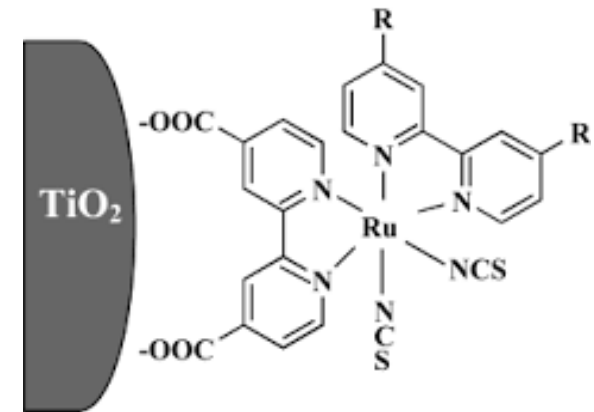
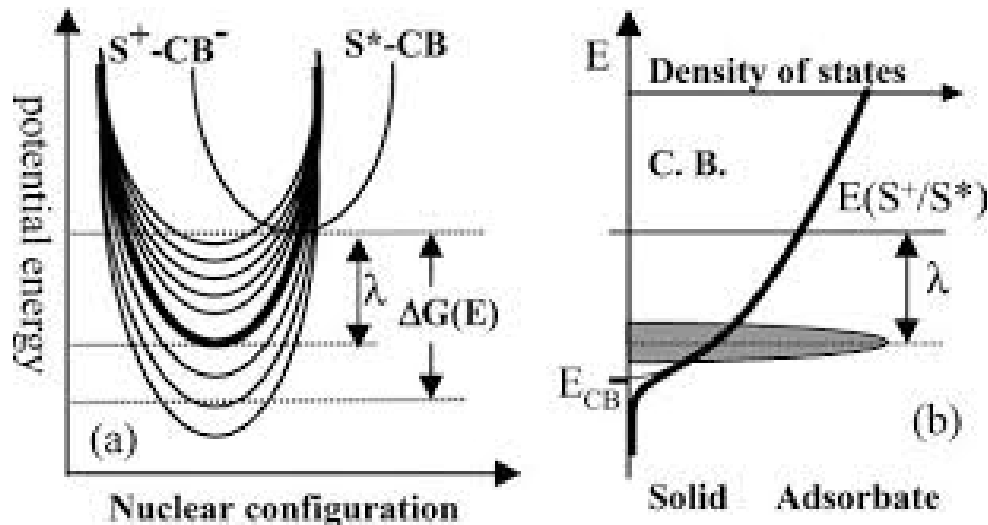
Marcus Theory of Electron Transfer

Extensions to the original Marcus theory:

- ***strong electronic coupling: Marcus-Hush theory***

the inner sphere electron transfer theory of Noel S. Hush refers to a continuous change of the electron density during transfer along a geometrical coordinate (adiabatic case).

- ***Interfacial Electron Transfer:***



N. A. Anderson, T. Lian, "Ultrafast electron transfer at the molecule-semiconductor nanoparticle interface", *Annu. Rev. Phys. Chem.* 2005. 56:491–519.

- ***Coupled Electron and Proton Transfer:***

S. Hammes-Schiffer, A. Stuchebrukhov, "Theory of Coupled Electron and Proton Transfer Reactions", *Chem. Rev.* 2010. 110, 6930..

to be continued ...

